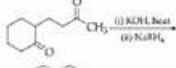



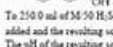


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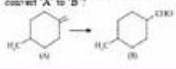
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- In which of the following species O.N. per atom of the underlined element is equal to +1?
 (a) $\text{S}_2\text{O}_7^{2-}$, Fe_2O_3 (b) $\text{S}_2\text{O}_7^{2-}$, N_2O
 (c) H_2PO_3 , Fe_2O_3 (d) N_2O , H_2PO_3
- In S_xI₂ reactions, the correct order of reactivity for the following compounds, CH₃Cl, CH₃CH₂Cl, (CH₃)₂CHCl and (CH₃)₃CCl is:
 (a) $(\text{CH}_3)_3\text{CCl} > (\text{CH}_3)_2\text{CHCl} > \text{CH}_3\text{CH}_2\text{Cl} > \text{CH}_3\text{Cl}$
 (b) $(\text{CH}_3)_3\text{CCl} > (\text{CH}_3)_2\text{CHCl} > \text{CH}_3\text{CH}_2\text{Cl} > \text{CH}_3\text{Cl}$
 (c) $\text{CH}_3\text{Cl} > (\text{CH}_3)_2\text{CHCl} > \text{CH}_3\text{CH}_2\text{Cl} > (\text{CH}_3)_3\text{CCl}$
 (d) $\text{CH}_3\text{Cl} > \text{CH}_3\text{CH}_2\text{Cl} > (\text{CH}_3)_2\text{CHCl} > (\text{CH}_3)_3\text{CCl}$
- The major product of the following reaction is:


 (a) 
 (b) 
 (c) 
 (d) 
- To 250.0 ml of 36.50 H₂SO₄, 4.0 g of solid NaOH is added and the resulting solution is diluted to 1.0 L. The pH of the resulting solution is
 (a) 12.0 (b) 11.25 (c) 11.95 (d) 12.95
- Match List-I with List-II

List-I	List-II
(a) $(\text{FeCl}_3)_2^{2+}$	(i) sp ³ d
(b) FeF_3	(ii) sp ² sp ²
(c) PCl_3	(iii) sp ³
(d) $(\text{Co}(\text{NH}_3)_6)^{3+}$	(iv) sp ³ d ²

Choose the most appropriate answer from the options given below:
 (a) (a)-(ii), (b)-(iv), (c)-(i), (d)-(iii)
 (b) (a)-(iii), (b)-(iv), (c)-(i), (d)-(ii)
 (c) (a)-(iii), (b)-(i), (c)-(iv), (d)-(ii)
 (d) (a)-(ii), (b)-(i), (c)-(iv), (d)-(iii)
- Which of the following isopropylbenzenes will convert 'A' to 'B'?



(a) PCC oxidation
 (b) Chromatography
 (c) H_2 , Pd/C , CHCl₃ followed by PCC oxidation
 (d) HBr, hydrolysis followed oxidation by $\text{K}_2\text{Cr}_2\text{O}_7$

- Solute A associates in water. When 0.7 g of solute A is dissolved in 42.0 g of water, it depresses the freezing point by 0.2 °C. The percentage association of solute A in water, is [Given: Molar mass of A = 93 g mol⁻¹, Molal depression constant of water is 1.86 K kg mol⁻¹]
 (a) 50% (b) 60% (c) 70% (d) 80%
- The correct set of the products obtained in the following reactions is:
 (A) $\text{RCN} \xrightarrow{\text{acid}}$
 (B) $\text{RCN} \xrightarrow{\text{H}_2\text{O}}$
 (C) $\text{RNC} \xrightarrow{\text{hydrolysis}}$
 (D) $\text{RNC} \xrightarrow{\text{H}_2\text{O}}$
 The answer is:
 (i) A: 2° Amine, B: Methyl ketone
 (ii) A: 1° Amine, B: Alcohol
 (iii) A: 2° Amine, B: Methyl ketone
 (iv) A: 2° Amine, B: Alcohol
 (v) A: 2° Amine, B: Methyl ketone
 (vi) A: 2° Amine, B: Alcohol
 (vii) A: 2° Amine, B: Methyl ketone
 (viii) A: 2° Amine, B: Alcohol
- If a compound on analysis was found to contain C = 18.5%, H = 1.55%, Cl = 55.04% and O = 24.81%, then its empirical formula is:
 (a) CHClO (b) CHClO
 (c) C₂H₂Cl₂O₂ (d) C₂H₂Cl₂O
- Assertion (A): Aniline on nitration yields ortho, meta & para nitro derivatives of aniline.
 Reason (R): Nitrating mixture is a strong acidic mixture.
- In lactose, the reducing part is:
 (a) Galactose (b) Glucose
 (c) Fructose (d) Mannose
- For any given series lines of atomic hydrogen, let $\Delta \bar{\nu} = \bar{\nu}_{max} - \bar{\nu}_{min}$ be the difference in maximum and minimum frequencies in m⁻¹. The ratio $\Delta \bar{\nu}_{series} / \Delta \bar{\nu}_{series}$ is:
 (a) 4 : 1 (b) 9 : 4 (c) 27 : 5 (d) 5 : 4
- Bond dissociation enthalpy of E-H (E = element) bonds is given below. Which of the compounds will act as strongest reducing agent?

Compound	NH ₃	PH ₃	AsH ₃	SbH ₃
$\Delta_{bond} (E-H) / \text{kJ mol}^{-1}$	389	322	297	255

(a) NH_3 (b) PH_3 (c) AsH_3 (d) SbH_3

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IBM Guardium Data Protection v12.x Administrator - Professional Sample Questions (Q63-Q68):

NEW QUESTION # 63

What is the underlying technology in the architecture of Guardium Universal Connector?

- A. Filebeat server
- B. Kafka cluster
- C. Kafka nodes
- D. Logstash pipeline

Answer: D

NEW QUESTION # 64

How can Guardium administrators demonstrate compliance improvements after applying hardening measures identified through CAS?

- A. By scheduling appliance firmware rollbacks
- B. By disabling all anomaly detection policies
- C. By comparing CAS scores before and after remediation
- D. By enabling S-TAP auto-reconnect to collectors

Answer: C

NEW QUESTION # 65

Which Guardium tool can administrators use to validate the integrity of data stored on appliances during health audits?

- A. Network discovery wizard
- B. Policy builder
- C. Data management utility
- D. S-TAP installer

Answer: C

NEW QUESTION # 66

What two outcomes can Guardium's sensitive data classification provide to help an organization strengthen compliance? (Choose two)

- A. Change database table structures to encrypt columns
- B. Automatically create central manager accounts
- C. Identify columns containing regulated data such as PII
- D. Produce classification reports for auditors

Answer: C,D

NEW QUESTION # 67

Where can data archived on an aggregator be restored?

- A. On any aggregator
- B. On any collector
- C. On a standalone collector
- D. On a collector that was exported to the aggregator

Answer: A

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