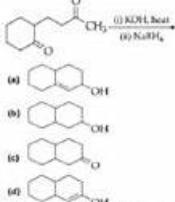
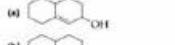
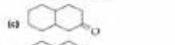
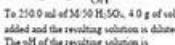
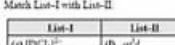


# C1000-197 Latest Exam Cost | C1000-197 Exam Questions And Answers

1. In which of the following species O:N per atom of the undefined element is/are equal to +1?  
 (a)  $\text{S}_2\text{O}_3^{2-}$ ,  $\text{P}_2\text{O}_7^{4-}$       (b)  $\text{P}_2\text{O}_7^{4-}$ ,  $\text{N}_2\text{O}$   
 (c)  $\text{H}_3\text{PO}_4$ ,  $\text{Fe}_2\text{O}_3$       (d)  $\text{N}_2\text{O}_4$ ,  $\text{H}_2\text{O}_2$

2. In  $\text{S}_{\text{N}}2$  reactions, the correct order of reactivity for the following compounds,  $\text{CH}_3\text{Cl}$ ,  $\text{CH}_3\text{CH}_2\text{Cl}$ ,  $(\text{CH}_3)_2\text{CHCl}$  and  $(\text{CH}_3)_3\text{CCl}$  is:  
 (a)  $(\text{CH}_3)_3\text{CCl} > (\text{CH}_3)_2\text{CHCl} > \text{CH}_3\text{CH}_2\text{Cl} > \text{CH}_3\text{Cl}$   
 (b)  $(\text{CH}_3)_2\text{CHCl} > (\text{CH}_3)_3\text{CCl} > \text{CH}_3\text{CH}_2\text{Cl} > \text{CH}_3\text{Cl}$   
 (c)  $\text{CH}_3\text{Cl} > (\text{CH}_3)_2\text{CHCl} > \text{CH}_3\text{CH}_2\text{Cl} > (\text{CH}_3)_3\text{CCl}$   
 (d)  $\text{CH}_3\text{Cl} > \text{CH}_3\text{CH}_2\text{Cl} > (\text{CH}_3)_3\text{CCl} > (\text{CH}_3)_2\text{CHCl}$

3. The major product of the following reaction is:  


(a)   
 (b)   
 (c)   
 (d) 

4. To 25.0 mL of 3M-50 H<sub>2</sub>SO<sub>4</sub>, 4.0 g of solid NaOH is added and the resulting solution is diluted to 1.0 L. The pH of the resulting solution is  
 (a) 12.0 (b) 11.25 (c) 11.95 (d) 12.95

5. Match List-I with List-II

List-I	List-II
(a) $[\text{PCl}_3]^{2-}$	(i) $\text{sp}^2$
(b) $\text{BrF}_3$	(ii) $\text{sp}^3$
(c) $\text{PCl}_5$	(iii) $\text{sp}^2$
(d) $[\text{Co}(\text{NH}_3)_6]^{2+}$	(iv) $\text{sp}^3$

Choose the most appropriate answer from the options given below:

(a) (a)-(II), (b)-(IV), (c)-(I), (d)-(III)  
 (b) (a)-(II), (b)-(IV), (c)-(I), (d)-(III)  
 (c) (a)-(II), (b)-(I), (c)-(IV), (d)-(II)  
 (d) (a)-(II), (b)-(I), (c)-(IV), (d)-(III)

6. Which of the following reagents/conditions will convert 'A' to 'B'?



64. PCC oxidation  
 (a) Oxidation  
 (b)  $\text{CH}_3\text{CO}_2\text{O}^{+}/\text{CH}_3\text{CO}^{+}$  followed by PCC oxidation  
 (c)  $\text{Hg}^{2+}$ , hydrolysis followed by oxidation by  $\text{K}_2\text{Cr}_2\text{O}_7$

7. Solute A dissociates in water. When 0.7 g of solute A is dissolved in 42.0 g of water, it depresses the freezing point by 0.2 °C. The percentage association of solute A in water, is [Given, Molar mass of A = 93 g/mol, Molar depression constant of water is 1.86 K kg mol<sup>-1</sup>]  
 (a) 50% (b) 60% (c) 70% (d) 80%

8. The correct set of the products obtained in the following reaction is:

(A)  $\text{RCH}_2 \xrightarrow{\text{H}_2\text{O}, \text{Hg}^{2+}}$   
 (B)  $\text{RCH}_2 \xrightarrow{\text{H}_2\text{O}, \text{Hg}^{2+}}$   
 (C)  $\text{RCH}_2 \xrightarrow{\text{H}_2\text{O}, \text{Hg}^{2+}}$   
 (D)  $\text{RCH}_2 \xrightarrow{\text{H}_2\text{O}, \text{Hg}^{2+}}$

The answer is:  
 (A) A  
 (B) Methyl ketone  
 (C) O  
 (D) A  
 (E) Alkene  
 (F) B  
 (G) C  
 (H) D  
 (I) E  
 (J) F  
 (K) G  
 (L) H  
 (M) I  
 (N) J  
 (O) K  
 (P) L  
 (Q) M  
 (R) N  
 (S) O  
 (T) P  
 (U) Q  
 (V) R  
 (W) S  
 (X) T  
 (Y) U  
 (Z) V

9. If a compound on analysis was found to contain C = 18.5%, H = 1.55%, Cl = 53.04% and O = 24.81%, then its empirical formula is

(a)  $\text{CHClO}$       (b)  $\text{CH}_2\text{ClO}$   
 (c)  $\text{C}_2\text{H}_2\text{Cl}_2$       (d)  $\text{CHCl}_2\text{O}$

10. Assertion (A) : Aniline on nitration yields ortho, meta & para nitro derivatives of aniline.  
 Reason (R) : Nitrating mixture is a strong acidic mixture

11. In lactose, the reducing part is  
 (a) Galactose      (b) Glucose  
 (c) Fructose      (d) Mannose

12. For any given series lines of atomic hydrogen, let  $\Delta V = \Delta V_{\text{max}} - \Delta V_{\text{min}}$  be the difference in maximum and minimum frequencies in  $\text{cm}^{-1}$ . The ratio  $\Delta V_{\text{max}}/\Delta V_{\text{min}}$  is:

(a) 4 : 1 (b) 9 : 4 (c) 27 : 5 (d) 5 : 4

13. Bond dissociation enthalpy of E - H (E = element) bonds is given below. Which of the compounds will act as strongest reducing agent?

Compound	$\text{NH}_3$	$\text{PH}_3$	$\text{AsH}_3$	$\text{BH}_3$
$\Delta_{\text{diss}}(E - H = \text{AsH}_3)$ (kJ mol <sup>-1</sup> )	359	322	297	255

(a)  $\text{NH}_3$       (b)  $\text{PH}_3$       (c)  $\text{AsH}_3$       (d)  $\text{BH}_3$

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## IBM Guardium Data Protection v12.x Administrator - Professional Sample Questions (Q63-Q68):

### NEW QUESTION # 63

What is the underlying technology in the architecture of Guardium Universal Connector?

- A. Filebeat server
- B. Kafka cluster
- C. Kafka nodes
- D. Logstash pipeline

**Answer: D**

### NEW QUESTION # 64

How can Guardium administrators demonstrate compliance improvements after applying hardening measures identified through CAS?

- A. By scheduling appliance firmware rollbacks
- B. By disabling all anomaly detection policies
- C. By comparing CAS scores before and after remediation
- D. By enabling S-TAP auto-reconnect to collectors

**Answer: C**

### NEW QUESTION # 65

Which Guardium tool can administrators use to validate the integrity of data stored on appliances during health audits?

- A. Network discovery wizard
- B. Policy builder
- C. Data management utility
- D. S-TAP installer

**Answer: C**

### NEW QUESTION # 66

What two outcomes can Guardium's sensitive data classification provide to help an organization strengthen compliance? (Choose two)

- A. Change database table structures to encrypt columns
- B. Automatically create central manager accounts
- C. Identify columns containing regulated data such as PII
- D. Produce classification reports for auditors

**Answer: C,D**

### NEW QUESTION # 67

Where can data archived on an aggregator be restored?

- A. On any aggregator
- B. On any collector
- C. On a standalone collector
- D. On a collector that was exported to the aggregator

**Answer: A**

## NEW QUESTION # 68

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