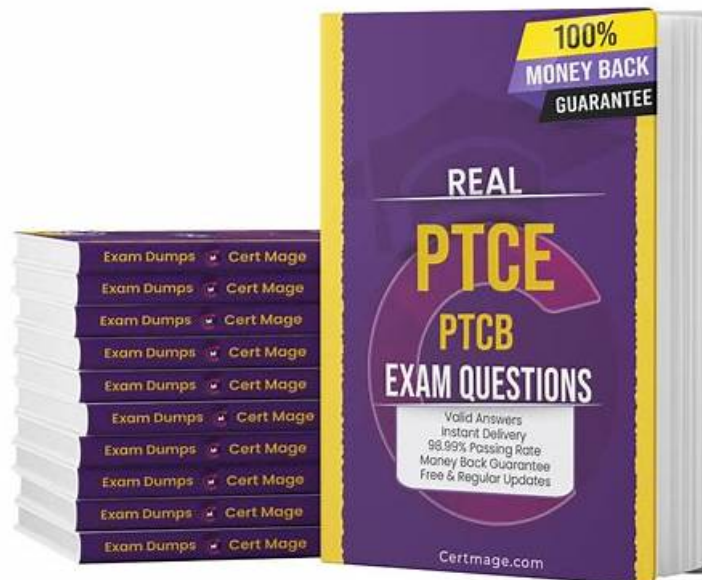


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Medical Tests Pharmacy Technician Certification Exam Sample Questions (Q242-Q247):

NEW QUESTION # 242

A patient weights 218 Lbs. What is their weight in Kg?

- A. 409Kg
- **B. 99Kg**
- C. 109Kg
- D. 479Kg

Answer: B

NEW QUESTION # 243

Which of the following medications is subject to a Risk Evaluation and Mitigation Strategy (REMS) program?

- A. Warfarin
- **B. Suboxone**
- C. Xarelto
- D. Haloperidol

Answer: B

Explanation:

Risk Evaluation and Mitigation Strategies (REMS) are FDA-required programs to ensure that the benefits of certain medications outweigh their risks. REMS programs restrict prescribing, dispensing, and monitoring due to the potential for serious side effects or misuse.

* Suboxone is used for opioid dependence and contains buprenorphine, a partial opioid agonist, which can cause addiction, misuse, and respiratory depression if used improperly.

* REMS Requirements for Suboxone:

* Prescriber Certification: Only specially certified providers can prescribe Suboxone.

* Patient Education: Patients must be informed about the risks of misuse, overdose, and dependence.

* Limited Dispensing: Pharmacists must ensure prescriptions meet REMS criteria before dispensing.

* A. Xarelto (Rivaroxaban).# Incorrect. While Xarelto is an anticoagulant, it is not under REMS.

* C. Warfarin.# Incorrect. Warfarin requires monitoring but does not have a REMS program.

* D. Haloperidol.# Incorrect. Haloperidol is an antipsychotic, but it does not have a REMS program.

* FDA REMS Program List (www.accessdata.fda.gov).

* Suboxone REMS Guidelines (www.suboxonerems.com).

* PTCB PTCE Exam Content Outline - REMS and Medication Safety.

* Mosby's Pharmacy Technician: Principles and Practice - REMS Regulations.

Why Is Suboxone (Buprenorphine/Naloxone) Under REMS? Why the Other Answer Choices Are Incorrect:

Pharmacy Technician References:

NEW QUESTION # 244

Incompatibility of a nonsterile compound may be indicated by:

- A. a clear solution
- B. the use of a pH buffer system
- **C. an unexpected change in odor**
- D. suspension particles settling to the bottom of a container

Answer: C

Explanation:

Incompatibility in a nonsterile compound can be identified through various physical, chemical, and therapeutic changes. An unexpected change in odor is a key indicator of chemical instability, contamination, or degradation of a compounded preparation. Some ingredients may break down into volatile compounds, leading to noticeable odor changes.

* A. A clear solution# Incorrect; clarity alone does not indicate incompatibility. Some solutions may be clear but still chemically unstable.

* B. The use of a pH buffer system# Incorrect; buffer systems are used to maintain pH stability and prevent incompatibility rather than indicate it.

* D. Suspension particles settling to the bottom of a container# Incorrect; suspensions naturally settle and require shaking before use. This is not necessarily a sign of incompatibility.

Other Answer Choices Explained: References:

* USP <795> Pharmaceutical Compounding-Nonsterile Preparations

* PTCB PTCE Exam Content Outline (Medication Safety & Quality Assurance)
* Remington: The Science and Practice of Pharmacy

NEW QUESTION # 245

How many milligrams are in 1 kilogram?

- A. 1,000,000
- B. 1,000
- C. 100,000
- D. 10,000

Answer: A

NEW QUESTION # 246

Dr. Diaco has prescribed your customer a drug order for the following: Methylprednisolone 10mg tabs
4 0mg QD x2D 30mg QD X3D 20mg QD X4D 10mg QD X5D
How many tablets should be dispensed?

- A. 0
- B. 1
- C. 2
- D. 3

Answer: D

NEW QUESTION # 247

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