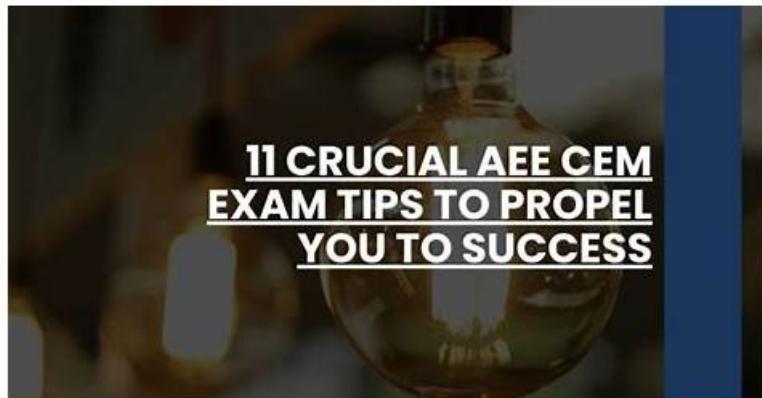


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AEE Certified Energy Manager (CEM) Sample Questions (Q29-Q34):

NEW QUESTION # 29

A compressed air system has a total volume of 247,500 liters. During a 30-minute break in production (air compressor system turned off and no production demand for compressed air), the system pressure drops from 7 bar, gauge (700 kPa, gauge) to 5 bar, gauge (500 kPa, gauge). Calculate the leakage rate in the compressed air system.

- A. 32,000 liters/minute
- B. 16,500 liters/minute
- C. 48,250 liters/minute
- D. 22,750 liters/minute
- E. 8,250 liters/minute

Answer: B

NEW QUESTION # 30

A facility uses 250,000 kWh in a 31-day billing period. The peak demand in the billing period is 1,200 kW. What is the facility partial load factor?

- A. 28%
- **B. 33%**
- C. 77%
- D. 45%
- E. 66%

Answer: B

Explanation:

A screenshot of a calculator AI-generated content may be incorrect.

Partial Load Factor (PLF) is calculated as:

$$PLF = \frac{\text{Total energy consumption (kWh)}}{\text{Peak demand (kW)} \times \text{Total hours}}$$

Step 1: Identify Given Data

- Total energy consumption: 250,000 kWh
- Peak demand: 1,200 kW

TestKing 24 hours/day = 744 hours

Step 2: Calculate PLF

$$PLF = \frac{250,000}{1,200 \times 744}$$

$$PLF = \frac{250,000}{892,800}$$

$$PLF = 0.28 = 33\%$$

Thus, the correct answer is B. 33%.

NEW QUESTION # 31

A cooling tower is delivering water to a chilled-water plant. The cooling tower outlet temperature has a 6°C approach temperature to the ambient air, wet-bulb temperature. The outside air conditions are 20°C dry bulb, 60% relative humidity. Using the psychrometric chart, at what temperature can the cooling tower deliver water to the chilled-water plant?

- A. 21°C
- B. 25°C
- **C. 17°C**
- D. 13°C

Answer: C

NEW QUESTION # 32

A controlled valve is full closed when it gets a signal of 4 mA. The valve is full open when it gets a signal of 20 mA. With a control signal of 8 mA, what is the valve position?

- A. 12.5% open
- B. 45% open
- C. 82% open
- **D. 25% open**
- E. 75% open

Answer: D

NEW QUESTION # 33

How much energy (kJ) is required to heat 2 kg of water from 25°C to 2 kg of water at 100°C? (The specific heat of water (C) is 4.2 kJ/kg·°C. Assume no system losses.)

- A. 840 kJ
- B. 420 kJ
- C. 630 kJ
- D. 510 kJ

Answer: A

NEW QUESTION # 34

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